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RJ Shavelson

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interatomic linkage that results
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pair between two atoms the
binding arises from the
electrostatic attraction of their
nuclei for the same electrons a
bond forms when the bonded
atoms have a lower total
energy than that of widely
separated atoms

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covalent nonpolar covalent the
greater the electronegativity
difference between two atoms
bonded together the greater
the bond s percentage of ionic
character nonpolar character
metallic character electron
sharing the electrons involved
in the formation of a chemical
bond are called valence
electrons

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molecular orbital the region of
high probability of finding an
electron that is being shared
between two nuclei in a
covalent bond bond length
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making it take more energy to
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